



UNIVERSITY OF NIŠ

Course Unit Descriptor

Faculty

Faculty of Mechanical Engineering

GENERAL INFORMATION

| | | | |
|------------------------------|---|--|---|
| Study Program | Mechanical Engineering | | |
| Study Module (if applicable) | - | | |
| Course Title | Thermodynamics | | |
| Level of Study | <input checked="" type="checkbox"/> Bachelor | <input type="checkbox"/> Master's | <input type="checkbox"/> Doctoral |
| Type of Course | <input checked="" type="checkbox"/> Obligatory | <input type="checkbox"/> Elective | |
| Semester | <input type="checkbox"/> Autumn | <input checked="" type="checkbox"/> Spring | |
| Year of Study | II | | |
| Number of ECTS Allocated | 7 | | |
| Name of Lecturer/Lecturers | Mića V. Vukić | | |
| Teaching Mode | <input checked="" type="checkbox"/> Lectures | <input type="checkbox"/> Group tutorials | <input type="checkbox"/> Individual tutorials |
| | <input checked="" type="checkbox"/> Laboratory work | <input type="checkbox"/> Project work | <input type="checkbox"/> Seminar |
| | <input type="checkbox"/> Distance learning | <input type="checkbox"/> Blended learning | <input type="checkbox"/> Other |

Purpose and Overview (max. 5 sentences)

Introduce students to the principles and limitations of thermal energy transformation. Practical applications. Mechanisms of heat transfer.

Syllabus (brief outline and summary of topics, max. 10 sentences)

1) Introductory concepts and definitions. Describing thermodynamic systems and their behaviour. Ideal gas model. (p–v–T) relation. 2) Ideal gas mixtures. (p–v–T) relation for ideal gas mixtures. 3) Energy and the first law of thermodynamics. Internal energy and specific heat of ideal gases. Work and heat. Energy transfer by heat. Energy balance for closed systems. (p,v) diagram. 4) Open thermodynamic systems. Kinetic energy. Potential energy. Enthalpy. Steady-state flow processes. Conservation of mass for a control volume and conservation of energy for a control volume. 5) Polytropic and other process of an ideal gas. Thermodynamic cycle. 6) Introducing entropy and the second law of thermodynamics. 7) (T,s) diagram. Defining entropy change. Heat transfer and work in internally reversible. 8) Isentropic processes. Isentropic efficiencies. 9) Real gasses. Steam power systems. Analyzing steam power systems – Rankin's cycle. Other cycles. Refrigeration and heat pump cycles. 10) Energy transfer. Conduction. Fourier's law. 11) Convection. Newton's law. 12) Thermal radiation. Stefan-Boltzmann's law.

Language of Instruction

- Serbian (complete course) English (complete course) Other _____ (complete course)
- Serbian with English mentoring Serbian with other mentoring _____

Assessment Methods and Criteria

| Pre exam Duties | Points | Final Exam | Points |
|---------------------------------|---------------|----------------------------|---|
| Activity During Lectures | 5 | Written Examination | 0 (or max 60 depending on Pre exam Duties) |
| Practical Teaching | 15 | Oral Examination | Max. 30 (depending on Teaching Colloquia) |
| Teaching Colloquia | 50 | Overall Sum | 100 |

***Final examination mark is formed in accordance with the Institutional documents**